

Assignment V, PHYS 230 (Introduction to Modern Physics)
Fall 2015
Due Thursday, 10/1/15 at start of class

This homework is fairly straightforward. Consider it the calm before the storm.

1. A double slit with slit separation d is placed in front of a red laser ($\lambda = 650$ nm). 4 meters away, you notice that the distance between the central maximum and the *second* minimum is 9 cm. (In other words, there's a total of two minima between the central maximum and the location of this minimum).
 - a) What is the slit separation?
 - b) If the screen was moved from 4 meters away to 7 meters away, how far would it be between the 6th minima to the left of the central maximum to the 3rd maximum to the right of the central maximum?
2. A diffraction grating with 2000 lines per cm is used to separate the light from a source that has wavelengths $\lambda_0 = 493$ nm and $\lambda_1 = 524$ nm. If the light goes through the grating and is viewed on a screen 2 meters away, what physical distance would there be between the first maxima of the two constituent wavelengths?
3. X-ray tubes used by dentists often accelerate electrons with a potential difference of about 80 kV. What is the minimum wavelength of the x-rays that are produced?
4. The smallest angle of Bragg scattering in potassium chloride (KCl) is 28.4° for 0.30 nm x-rays. Find the distance between atomic planes in potassium chloride.
5. The work function of Molybdenum is 4.22 eV.
 - a) What is the threshold frequency for the photoelectric effect in Molybdenum?
 - b) Will yellow light of wavelength 560 nm cause ejection of photoelectrons from Molybdenum? Prove your answer.
6. A photoelectric experiment with Cesium yields stopping potentials for $\lambda = 435.8$ nm and $\lambda = 546.1$ nm to be 0.95 V and 0.38 V, respectively. Using these data only, find the threshold frequency and work function for Cesium and the value of h .

More on Back!

7. One of the funky things that didn't follow the classical expectation for the photoelectric effect is that there was no time-lag between turning on the light-source and measuring a current. In practice – if we don't know about or believe the quantum hypothesis – then we would expect there to be some finite amount of time between when you turn on the source and when a typical electron could gain enough energy from the light beam to be liberated. Let's try and ballpark this expected time-lag for a weak source.
- a) Assume a lightbulb emits total power P equally in all directions. Let us put a metal surface X meters away from this light source, and let's assume it takes energy ϕ to liberate an electron from an atom in this metal. Assuming an atom has a circular cross-section of D , how long would it take for the atom to gain energy ϕ from the source? Leave your answer in terms of P , X , ϕ , and D . (Hint: To check your answer, make sure that it makes sense. The larger P is, the less time it should take. The larger X is, the longer it should take. The higher ϕ is, the longer it should take, and the larger D is, the shorter it should take. That should tell you something about the form of the answer.)
 - b) Find the actual value of the time-lag as designed in part *a* if P is 2 W, X is 0.1 m, ϕ is 6 eV, and D is 0.1 nm.
 - c) Assuming that the photon hypothesis is correct (and that photons travel at c), how long would you expect the time-lag to be? Again, assume that $\phi = 6$ eV, that the wavelength of the light is 100 nm, and assume that the path the electron takes through the tube is a straight-line path (no acceleration once liberated). You may ignore relativistic effects for the electron.