

Assignment VII, PHYS 308
Fall 2014
Due 10/17/14 at start of class

NOTE: Please leave your answers in terms of actual numbers (with appropriate units) when possible. Please provide full, legible, easy to follow solutions to the following problems. I can't give you credit if I can't follow your reasoning. Extensive exposition on your thought process or strategy is always appreciated.

1. I gave you (in class) the following expression for the Diffusion Constant of an aerosol particle: $D = \frac{kTCC}{3\pi\mu D_p}$. (Note that this expression has units of m²/s). Another route to the diffusion equation developed from kinetic theory gives:

$$D^* = \frac{\lambda^2}{2t_{\text{ave}}}$$

where λ is the mean free path between collisions for a particle, and t_{ave} is the mean time between collisions. In practice, both of these expressions are legitimate ways of writing the diffusion constant. The first method makes a bit more sense for aerosols (where there is a particle and a fluid medium that are clearly different from each other). Using D^* makes more sense if you're just talking about diffusion of a single gas into, say, a vacuum or something. This problem is about playing with the two different definitions of D simultaneously.

- a) For methane, $D = 1.78 \times 10^{-5}$ m²/s. Assume $\lambda = 90$ nm. Find the mean time between intermolecular collisions for methane.
- b) What is the mean *instantaneous* speed for a methane molecule? (Hint....speed = distance/time).
- c) What is the root mean squared displacement from the origin for a methane molecule released at $t = 0$ at the origin and left to travel for 3 minutes?
- d) What is the actual distance that the methane molecule moved in this time? (Not the displacement, but the distance).
- e) Derive an expression for (root mean squared displacement from the origin)/(distance traveled) as a function of t . Leave your answer in terms of D , t , and v_{ave} (the average instantaneous velocity of a methane molecule).
- f) It is physically impossible for the diffusion distance to be larger than the distance traveled. This means that the expression you calculated in part (e) only can hold when $t > t_o$. Find the smallest possible value of t_o for methane.

2. Under certain circumstances, the radiation inside the sun can be treated like a gas of photons with the mean free path between interactions nominally 1 mm. (We say interactions instead of collisions because we're really talking about absorption and re-emission here. Instead of "colliding" with other photons, a photon moves for nominally 1 mm before being absorbed by something. Then it is re-emitted. You may assume this absorption/re-emission process is instantaneous). (This is a rather crude model, but it works for some situations).
- Under this basic model, how long (on average) would it take a photon created at the center of the sun to diffuse out to the surface of the sun? (It is not, of course, the "same" photon – but let's say we're tracing the energy path instead of the photon's path. We're abusing language a bit, but the language is at least evocative). Leave your answer in years.
 - What is the ratio:
(root mean squared displacement from the center of the sun)/(distance traveled)
just as the photon gets to the sun's surface?
3. In class, we developed the following expression for G_{rel} :

$$G_{\text{rel}} = 4\pi r^2 \sigma - \frac{4}{3}\pi r^3 nkT \ln\left(\frac{e}{e_s}\right)$$

- Find the value of r when G_{rel} reaches a maximum. (We call this value r_c for the critical radius). You may assume all parameters in the problem except for r can be treated as constants. You may also assume $e > e_s$.
- Find the value of G_{rel} when $r = r_c$. Please simplify your answer as much as you reasonably can.
- For $e > e_s$, $G_{\text{rel}} = 0$ at two values of r . The first is at $r = 0$. Find the other value of r that makes this true. We'll refer to the value of r you find here as $r_{2\text{nd zero}}$ (in the following parts of this problem).
- Let's compare $r_{2\text{nd zero}}$ to r_c . You might expect $r_{2\text{nd zero}} = 2r_c$ if the curve is completely symmetric for positive values of G_{rel} . Is $r_{2\text{nd zero}}$ greater than, equal to, or less than $2r_c$?
- How many liquid water molecules would be in a sphere of radius $r_{2\text{nd zero}}$? Assume $e/e_s = 1.042$, $\sigma = 0.076 \text{ J/m}^2$, and $T = 270\text{K}$.
- Technically, anything with $r > r_{2\text{nd zero}}$ has $G_{\text{rel}} < 0$ and thus is energetically favorable to exist compared to vapor. However, we do not see homogeneous nucleation of water drops that start at $r_{2\text{nd zero}}$ and grow. Explain why we don't see this.