## Assignment VIII, PHYS 230 (Introduction to Modern Physics) Fall 2015 Due Thursday, 10/22/15 at start of class

- 1. Numerically approximate the de Broglie wavelengths of the following:
  - a) An electron with kinetic energy 13.6 eV.
  - b) The fastest human being ever recorded to run (instantaneous velocity). (Google some stuff if necessary).
  - c) The Earth. (You may assume the sun is stationary. Again, you may have to Google some stuff).
  - d) An average Sodium molecule in an ideal gas at 500 pK. (The average speed of a molecule in an ideal gas can be computed via  $v_{\text{avg}} = \left(\frac{8kT}{\pi m}\right)^{1/2}$  with k the Boltzmann constant. A picoKelvin is  $10^{-12}$  K). (Think carefully about what mass to use).
  - e) Compare your answer in part (d) to the accepted value of Sodium's atomic radius (approximately 186 picometers). Briefly comment.
- 2. The formulation of much of the wave mechanics that will appear in the coming weeks relies on the formulation of a "wave equation". The most familiar wave equation is a differential equation of the form:

$$\frac{\partial^2 u}{\partial x^2} = \frac{1}{v^2} \frac{\partial^2 u}{\partial t^2}$$

where v is the velocity of the wave, and u(x,t) is a function describing the thing that's waving. Although many people in this class have not had differential equations – so "solving" this isn't necessarily fair for me to ask....there *are* things we can do with this equation that might help us understanding what's going on a bit better.

- a) Verify (by explicitly taking the necessary derivatives) that  $u = A\cos(kx \omega t)$  is a solution to this equation. (In doing this, you will find a relationship that must be obeyed between k,  $\omega$ , and v for this to be a valid solution. What is that relationship?) Treat A, k, and  $\omega$  as real constants.
- b) Verify (by explicitly taking the necessary derivatives) that  $u = Be^{i(kx-\omega t)}$  is a solution to this equation as well. (Again, you'll find the same relationship between  $k, \omega$ , and v.) Treat B, k, and  $\omega$  as real constants.

(More on back)

3. Two waves travel simultaneously along a long wire. Their wave functions (the solutions to the wave equation above) are:

$$u_1(x,t) = (0.002 \text{ m})\cos\left(\left[8.0 \frac{1}{\text{m}}\right]x - \left[400\frac{1}{\text{s}}\right]t\right)$$
$$u_2(x,t) = (0.002 \text{ m})\cos\left(\left[7.6 \frac{1}{\text{m}}\right]x - \left[380\frac{1}{\text{s}}\right]t\right)$$

where u and x are in meters and t is in seconds.

- a) Add these two waves together to form a single equation for  $u_1(x,t) + u_2(x,t)$ . Leave your answer in the general form  $U(x,t) = 2A \operatorname{trig}(\operatorname{mess}_1) \operatorname{trig}(\operatorname{mess}_2)$  with each "trig" indicating a sine or a cosine (sort of similar to equation 4.19 in your text, but with the messes involving time as well).
- b) What is the phase velocity of the resultant wave?
- c) What is the group velocity of the resultant wave?
- d) Calculate the spatial interval  $\Delta x$  between successive zeros of the group (packet) and relate it to  $\Delta k$ .
- 4. Let a particle of mass m be constrained to be between points -a/2 and +a/2 on the x-axis.
  - a) What is the minimum uncertainty in the particle's momentum?
  - b) What is the minimum uncertainty in the particle's kinetic energy? (You may ignore relativistic effects and assume there is no uncertainty in the particle's mass).
  - c) Using your result from (b) above, calculate the minimum energy of an electron between -a/2 and a/2 when  $a \sim 5.3 \times 10^{-11}$  m. (This distance is known as the "Bohr radius" and corresponds to the most likely distance between the proton and electron in a Hydrogen atom in its ground state).
  - d) Using your result from (b) above, calculate the minimum energy of an electron confined between -a/2 and a/2 when a = 0.01 m.
  - e) Using your result from (b) above, calculate the minimum energy of a 100 mg bead moving on a thin (frictionless) wire between two rigid stops that are 2 cm apart.